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For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most

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The molecular structure of carbon dioxide is one carbon atom with two oxygen atoms on opposite sides of it.

\_\_\_\_\_ 13. Covalent bonds exist when combining atoms give up or accept electrons. \_\_\_\_\_ 14. A molecule contains two atoms. Part C Matching Match each description in Column B to the correct term in Column A.



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Radioactive decay a. is a spontaneous process. b. occurs when an unstable nucleus loses energy. c. does not require any input of energy. d. all of the above

\_\_\_\_ 13. Which of the following types of

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radiation has no mass and no charge? a. alpha c. gamma b. beta d. positron \_\_\_\_\_

14. When U-238 undergoes radioactive decay by losing an alpha particle,

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Write your answer in the space provided.

Date Class 11. List the five traditional branches of chemistry and explain what each involves. 12. Discuss the role of chemistry in one of these areas: Materials, Energy, Medicine and



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Biotechnology, Agriculture, the Environment, or the Universe. 13. looks unhealthy and that, perhaps, the grass is dying.

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